

MUGBERIA GANGADHAR MAHAVIDYALAYA

P.O.—BHUPATINAGAR, Dist.—PURBA MEDINIPUR, PIN.—721425, WEST BENGAL, INDIA
NAAC Re-Accredited B+Level Govt. aided College
CPE (Under UGC XII Plan) & NCTE Approved Institutions
DBT Star College Scheme Award Recipient

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SYLLABUS

DEPARTMENT: CHEMISTRY

Mugberia Gangadhar Mahavidyalaya, Bhupatinagar, Purba Medinipur

Semester	Paper	Topic
SEM-I	C1T (Organic Chemistry-I)	Basics of Organic Chemistry –
		1.Bonding and Physical Properties
		2.General Treatment of Reaction Mechanism
		Stereochemistry – 1 (total)
	C2T (Physical	Chemical Thermodynamics
	Chemistry-I)	1. Kinetic Theory and Gaseous State
		2. Chemical Kinetics
	C1P (Organic Chemistry-I lab)	1. Separation of organic compound using solubility.
		2. Boiling point of organic liquid compound.
		3. Identification of a pure organic compound.
	C2P (physical Chemistry-I lab)	1. Heat of neutralization of a strong acid by a
	,	strong base.
		2. Study of kinetics of decomposition of H2O2
		3. Heat of solution of oxalic acid from solubility
		measurement.
		1.PH of unknown Buffer Solution .
		2. Study of kinetics of decomposition of H ₂ O ₂ .
	GE-1T	Atomic Structure, Chemical Periodicity, Acids and
		Bases, Redox reactions, General Organic
		Chemistry & Aliphatic Hydrocarbons
		1. Atomic structure 2. Chemical periodicity 3. Acid
		and bases 4. Redox reactions
		1. Fundamentals of organic chemistry 2.
		Stereochemistry 3. Nucleophilic substitution and
		Elimination Reactions 4. Aliphatic Hydrocarbons
	GE-1P	1. Estimation of sodium carbonate and sodium
		hydrogen carbonate.
		2. Estimation oxalic acid by KMno4.
		3. Estimation of water of crystallization in Mohr's
		salt by KMnO ₄ .
		4. Estimation of Fe(II) by K ₂ Cr ₂ O ₇ .
		5. Estimation of Cu(II) by Na ₂ S ₂ O ₃ .
		1.Detection special element
		2. Detection of functional groups
SEM II	C3T (Inorganic Chemistry-I)	1. Extra nuclear structure of atom
	- 1	2. Redox reactions and precipitation reactions.
		1. Chemical Periodicity 2. Acid – Base reactions.

	COD (T	I to the transfer of
	C3P (Inorganic Chemistry-I lab)	1. Acid and Base Titrations
		2. Oxidation –Reduction Titrimetric
	C4T (Organic Chemistry-II)	Stereochemistry II
		1.General Treatment of reaction Mechanism II
		2. Substitution and Elimination reactions
	C4P (Organic Chemistry-II lab)	Organic Preparations
	GE-2T	States of Matter & Chemical Kinetics, Chemical
		Bonding and Molecular Structure, P-block
		elements
		Kinetic theory of gases and real gases
		1. Liquids 2. Solids 3. Chemical kinetics
		Chemical Bonding and Molecular Structure
		2. Comparative study of P-block elements
	GE-2P	1. Surface tension measurements.
		2. Viscosity measurements. 3. Kinetics Study
		Qualitative study of Acid and Basic Radicals
SEM III	C5T	1. Viscosity 2. Chemical Equilibrium 3. Partial
	(Physical Chemistry-II)	properties and Chemical Potential
		1. Conductance and transport number.
		2. Fick's law 3. Foundation of Quantum
		Mechanics.
		Viscosity measurements.
		2. Determination of Partition Coefficient.
	C5P (Physical Chemistry-II lab)	3. Determination of equilibrium constant using
	(1 hysical chemistry-11 lab)	partition coefficient.
		1. Conductometric titration. 2. Study of
		saponification. 3. Verification of Ostwald,s
		dilution law.
	C6T (In angania Chamistay II)	
	C6T (Inorganic Chemistry-II)	Chemical Bonding -I
	C6P	Chemical Bonding –II. 2. Radioactivity I. Iodimetric Titrations .
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	(Inorganic Chemistry-II lab)	2. Estimation of metal content in some selective
		samples (Brass, Steel, Cement)
	CAT	1 Chamber of all 1 11
	C7T	1. Chemistry of alkenes and alkynes.
	(Organic Chemistry-III)	2. Aromatic Substitution.
		1. Carbonyl and Related Compounds
		2. Organometallics
	C7P (Organic Chemistry-III lab)	Qualitative analysis of single solid organic
		compounds
	SEC1T	Pharmaceutical Chemistry
		1. Drugs and Pharmaceuticals.
		2. Fermentation.
	SEC1P	Preparation of Aspirin and its analysis.
	SECII	Preparation of Aspirin and its analysis. Preparation of magnesium bisilicate .
-	CE 2T	1 5
	GE -3T	Chemical Energetics, Equilibria, Organic
		Chemistry-II
		1. Chemical Equilibrium.
		2. Ionic Equilibria.
		1. Aromatic Hydrocarbons 2. Organometallic

		Compounds 3. Aryl Halides 4. Alcohols, phenols,
	GE-3P	ether 5. Carbonyl Compounds.
	GE-SP	1. Determination of heat capacity of Calorimeter
		for different volumes. 2. Determination of enthalpy
		of ionization. 3. Determination of enthalpy of neutralization.
		1. Find the PH of an unknown buffer solution.
		2. Study of the solubility of benzoic acid in water.
SEM - IV	C8T	Study of the solubility of belizoic acid in water. Electromotive Force. 2. Quantum Chemistry.
SEWI-IV	(Physical Chemistry-III)	1. Colligative Properties 2. Phase rule 3. Binary
	(Filysical Chemistry-III)	solutions. 4. Ionic equilibria.
	C.8P	Determination of solubility of sparingly soluble
	(Physical Chemistry-III lab)	salt in water. 2. Determination of solubility product
	(1 hysical Chemistry-III lab)	3. Effect of ionic strength on the rate of
		Persulphate – Iodide reaction.
		1. Potentiometric titration of Mohr's salt .
		2. Study of phenol – Water phase diagram.
		3. PH - metric titration of acid against base.
	C.9T (Inorganic Chemistry-III)	General Principle of Metallurgy 2. Inorganic
		Polymers 3. Coordination Chemistry – 1.
		1. Chemistry of s and p Block elements .2. Nobel
		Gases.
	C9P	1.Complexometric titration
	(Inorganic Chemistry-III lab)	2.Inorganic Preparation
	G10F	177
	C10T	1.Nitrogens Compound
	(Organic Chemistry-IV)	2.Rearrangements
		1.Organic Synthesis 2.Organic Spectroscopy
	C10P	Quantitative Estimations
	(Organic Chemistry-IV lab)	Qualitative Estimations
	SEC -2T	Pesticides Chemistry
	SEC -2P	Determination of PH, acidity, alkalinity, ion
		exchange capacity. TLC, ion exchange paper
		chromatography method, Complexometric titration.
		Or
		Preparation
	GE4T	Solutions, Phase equilibria, Conductance,
		Electrochemistry and Analytical and Environmental
		Chemistry-I
		Solutions and Phase equilibrium
		Conductance and Electrochemistry
		Analytical and Environmental Chemistry
	GE4 P	Distribution law , Phase equilibria
	GLT I	Conductance, Potentiometry titration
SEM -V	C11T	Coordination Chemistry-II
~221,1	(Inorganic Chemistry-IV)	Chemistry of d- and f- block elements
	C11P	Chromatography of metal ions, Gravimetry and
	(Inorganic Chemistry-IV lab)	spectrophotometric method.
	C12T	Carbocycles and Heterocycles, Cyclic
	(Organic Chemistry-V)	Stereochemistry ,Pericyclic reactions
	(Signific Chemistry 1)	2001 De la companie d

		Carbohydrates, Bio-molecules
	C12P	Chromatographic Separations ,Spectroscopic
	(Organic Chemistry-V lab)	Analysis of Organic Compounds
	DSE-1T	Advanced Physical Chemistry
	DSE-11	
		Crystal Structure, Polymers
	DGE 1D	Statistical Thermodynamics
	DSE-1P	Computer programs based on numerical methods
	DSE-2T	Analytical Male de in Chamistan
		Methods in Chemistry
		Or
		Instrumental Methods
	DGE AD	of Chemical Analysis
	DSE-2P	Separation Techniques, Solvent Extractions,
		Spectrophotometry
		Or
GEN (I II	CIAT	Instrumental Methods of Chemical Analysis
SEM-VI	C13T	Bioinorganic Chemistry, Reaction Kinetics and
	(Inorganic Chemistry-V)	Mechanism
		Organometallic Chemistry
	C13P (Inorganic Chemistry-V	Qualitative Analysis
	lab	
	C14T	Molecular Spectroscopy, Surface phenomenon
	(Physical Chemistry-IV)	Photochemistry
	C14P	Determination of surface tension and CMC
	(Physical Chemistry-IV lab)	Verification of Beer and Lambert's Law, Study of
		kinetics of K2S2O8 + KI reaction, Determination
		of pH of unknown buffer and CMC
		spectrophotometrically
	DSE3T	Inorganic Materials
		of Industrial Importance
	DSE3P	1. Determination of free acidity in ammonium
		sulphate fertilizer.
		2. Estimation of Calcium in Calcium ammonium
		nitrate fertilizer.
		3. Estimation of phosphoric acid in superphosphate
		fertilizer.
		4. Electrolysis metallic coatings on ceramic and
		plastic material.
		5. Determination of composition of dolomite (by
		complexometric titration).
		6. Analysis of (Cu, Ni); (Cu, Zn) in alloy or
		synthetic samples.
		7. Analysis of Cement.
		8. Preparation of pigment (zinc oxide).
	DSE4T	Polymer Chemistry
	DSE4P	Polymer characterization, Polymer analysis.

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